

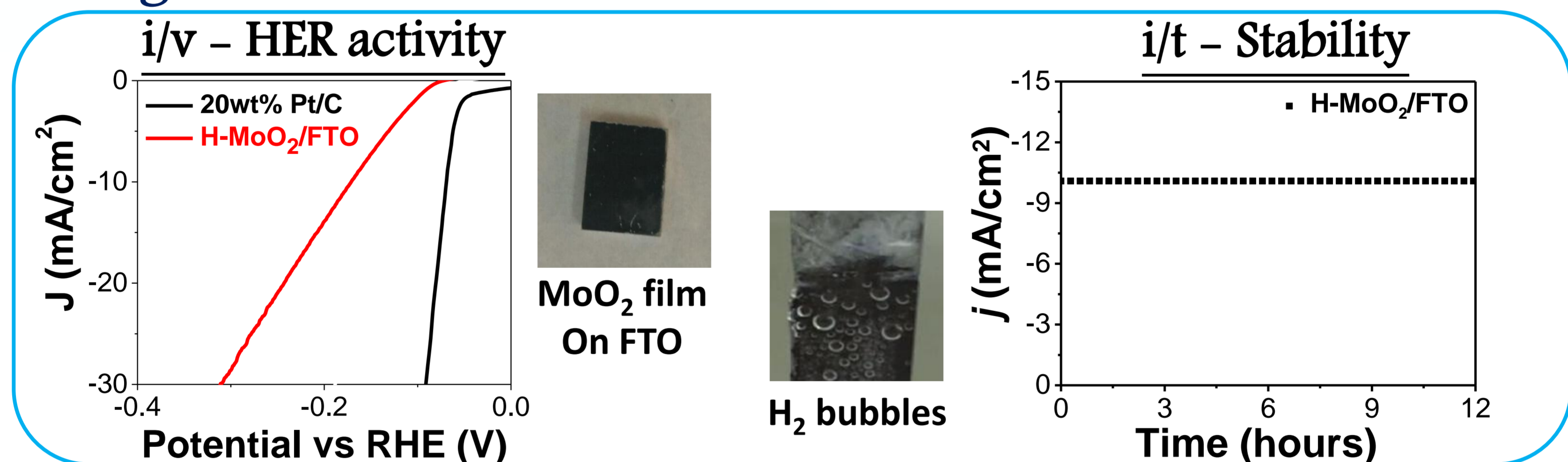
# Highly 'Robust' MoO<sub>2</sub> Nanostructures for *Electrochemical Hydrogen Generation*



- Immense energy demand & the need for sustainable and clean sources have driven Hydrogen (H<sub>2</sub>) production & storage in a cost-effective way.
- High cost and paucity of Pt – the bench mark catalyst
- MoO<sub>2</sub> nanostructures as alternate catalysts for hydrogen evolution reaction (HER)

## MoO<sub>2</sub> system

- ❖ Hydrogen annealed MoO<sub>2</sub> (H-MoO<sub>2</sub>) grown as film on fluorine-doped tin oxide (FTO) substrates
- ❖ Low onset potential of 72 mV demonstrated with 3 electrode configuration



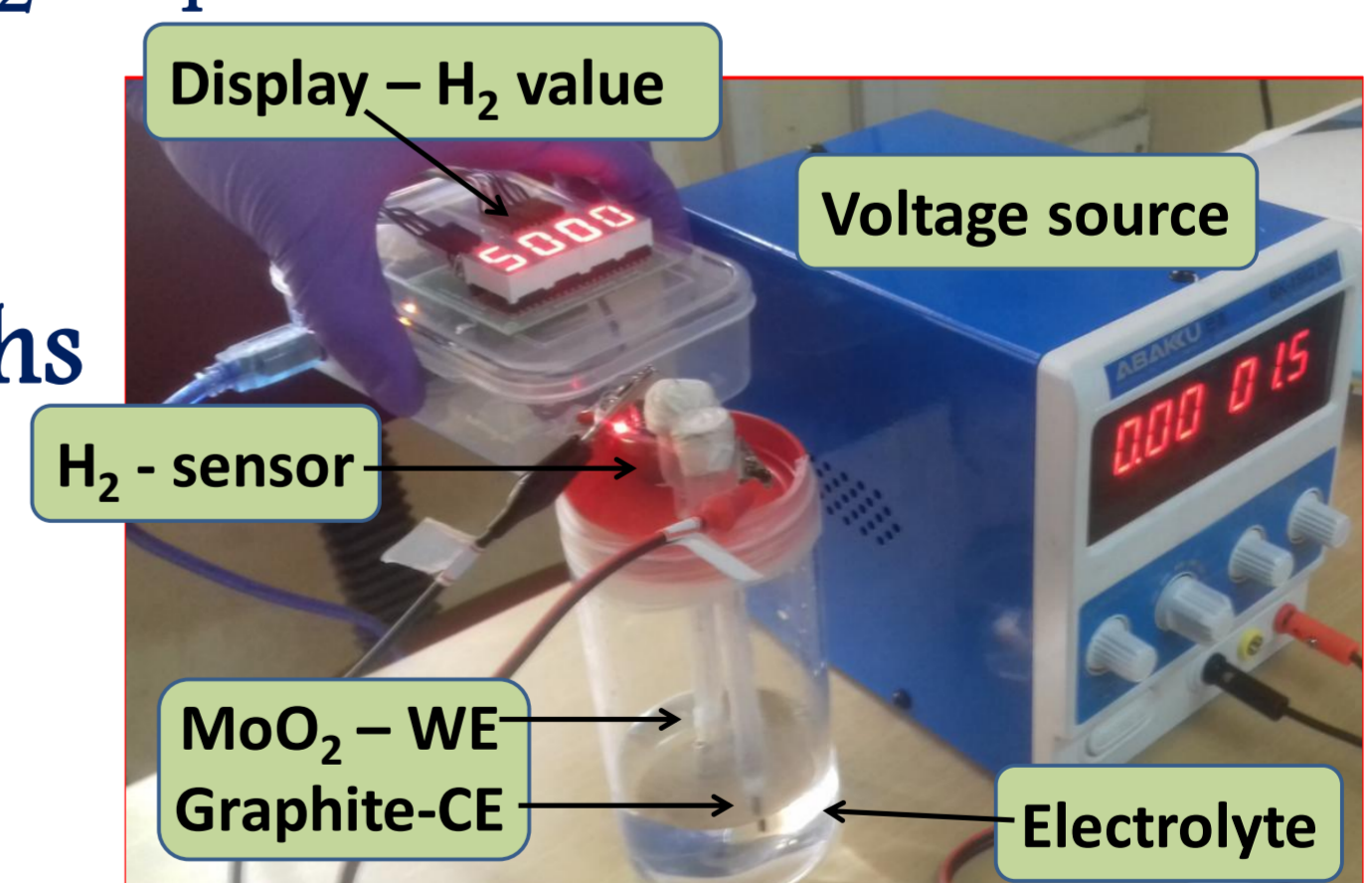
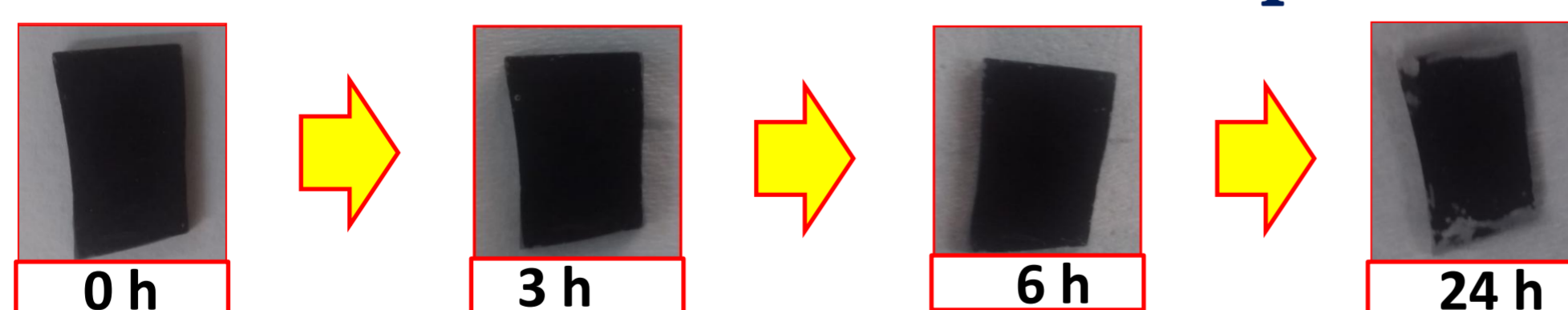
## Demo System

Faradaic efficiency of H<sub>2</sub> gas: 82% @ 10 mA/cm<sup>2</sup>

- ✓ 2 Electrode system – MoO<sub>2</sub>/FTO as cathode; graphite rod as anode
- ✓ Electrolyte: 0.5 M H<sub>2</sub>SO<sub>4</sub>; H<sub>2</sub> gas sensor (MQ-8); range: 10–10000 ppm.

## Highlight – Excellent stability

- ❖ Sonication test with bath sonicator in 0.5 M H<sub>2</sub>SO<sub>4</sub>
- ❖ 100% Stable upto 6h; after 24 hr ≥90%
- ❖ Electrochemical – several thousands of cycles
- ❖ Ambient – stable & active at least upto 6 months



## Comparison with Commercial Catalyst

Catalyst	Electrolyte	Onset	Overpotential @10 mA/cm <sup>2</sup>	Reference
20 wt% Pt/C	0.5 M H <sub>2</sub> SO <sub>4</sub>	37 mV	69 mV	<i>Chem. Eur. J.</i> , 2018,
H-MoO <sub>2</sub> /FTO	0.5 M H <sub>2</sub> SO <sub>4</sub>	72 mV	171 mV	doi:10.1002/chem.201803570

## Contributors:

Vivek Ramakrishnan and Neena S. John